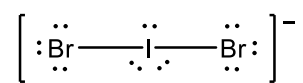


Draw IBr_2^- Lewis structure.

The Lewis structure for IBr_2^- is:



Iodine, a non-metallic element, has a total of 7 valence electrons. One of these electrons is used to form a single bond with bromine because the second one is a dative covalent bond.