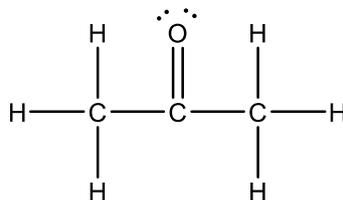


What is $(CH_3)_2CO$ Lewis structure?

The Lewis structure of $(CH_3)_2CO$ is:



Each carbon atom has four valence electrons, which are used to form chemical bonds.

Oxygen has six valence electrons, two of them are used to form a double bond with carbon while the other electrons represent two lone electronic pairs.