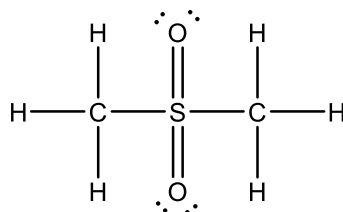


What is $CH_3SO_2CH_3$ Lewis structure?

The Lewis structure for $CH_3SO_2CH_3$ is:



Sulfur has six valence electrons which are used to form two double bonds with oxygen atoms and two single bonds with carbon atoms.

Carbon has four valence electrons which are used to form four single bonds, three with hydrogen atoms and one with sulfur.

Oxygen has six valence electrons, two are used to form one double bond with sulfur. The other four electrons represent two lone electronic bonds.