



What is $\text{CO}(\text{NH}_2)_2$ molar mass?

The molar mass of ethanol $\text{CO}(\text{NH}_2)_2$ is 60 g/mol.

Urea is a molecular compound with the chemical formula $\text{CO}(\text{NH}_2)_2$.

The molar mass of $\text{CO}(\text{NH}_2)_2$ is calculated:

$$\begin{aligned}M &= M_C + M_O + 2 \cdot M_N + 4 \cdot M_H \\M &= 12 \frac{\text{g}}{\text{mol}} + 16 \frac{\text{g}}{\text{mol}} + 2 \cdot 14 \frac{\text{g}}{\text{mol}} + 4 \cdot 1 \frac{\text{g}}{\text{mol}} \\M &= 60 \frac{\text{g}}{\text{mol}}\end{aligned}$$