



Which of the following occurs in an ionic bond?

- A. Oppositely charged ions attract
- B. Two atoms share two electrons
- C. Two atoms share more than two electrons
- D. Like-charged ions attract

In an ionic bond, there is a complete transfer of electrons from one atom to another atom, resulting in the formation of positively charged cations and negatively charged anions. The opposite charges attract each other and hold the ions together in a crystal lattice structure.

Therefore, the correct option is A. Oppositely charged ions attract